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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
09/961,395	09/25/2001	Harald Jakob	P 265258 000345 PV	5500
909 7	590 01/04/2005		EXAM	INER
PILLSBURY WINTHROP, LLP P.O. BOX 10500			LISH, P	ETER J
MCLEAN, VA 22102			ART UNIT	PAPER NUMBER
•			1754	

DATE MAILED: 01/04/2005

Please find below and/or attached an Office communication concerning this application or proceeding.

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	Application No.	Applicant(s)
	09/961,395	JAKOB ET AL.
Office Action Summary	Examiner	Art Unit
<u> </u>	Peter J Lish	1754
The MAILING DATE of this communication a Period for Reply	ppears on the cover sheet w	ith the correspondence address
A SHORTENED STATUTORY PERIOD FOR REP THE MAILING DATE OF THIS COMMUNICATION - Extensions of time may be available under the provisions of 37 CFR after SIX (6) MONTHS from the mailing date of this communication. - If the period for reply specified above is less than thirty (30) days, a re - If NO period for reply is specified above, the maximum statutory perio - Failure to reply within the set or extended period for reply will, by state Any reply received by the Office later than three months after the mai earned patent term adjustment. See 37 CFR 1.704(b).	1. 1.136(a). In no event, however, may a seply within the statutory minimum of thind will apply and will expire SIX (6) MO tute, cause the application to become A	reply be timely filed rty (30) days will be considered timely. NTHS from the mailing date of this communication. BANDONED (35 U.S.C. § 133).
Status		
 1) Responsive to communication(s) filed on 20 2a) This action is FINAL. 2b) The 3 This action is FINAL. 2b The 3 This action is application is in condition for allow closed in accordance with the practice under 	nis action is non-final. vance except for formal mat	·
Disposition of Claims		
4) ☐ Claim(s) 1-5,7-9 and 11-19 is/are pending in 4a) Of the above claim(s) is/are withden 5) ☐ Claim(s) is/are allowed. 6) ☐ Claim(s) 1-5, 7-9, and 11-19 is/are rejected. 7) ☐ Claim(s) is/are objected to. 8) ☐ Claim(s) are subject to restriction and	rawn from consideration.	
Application Papers		
9) The specification is objected to by the Examination The drawing(s) filed on is/are: a) and applicant may not request that any objection to the Replacement drawing sheet(s) including the correction. The oath or declaration is objected to by the	ccepted or b) objected to ne drawing(s) be held in abeya ection is required if the drawing	nce. See 37 CFR 1.85(a). g(s) is objected to. See 37 CFR 1.121(d).
Priority under 35 U.S.C. § 119		
12) Acknowledgment is made of a claim for foreign a) All b) Some * c) None of: 1. Certified copies of the priority docume 2. Certified copies of the priority docume 3. Copies of the certified copies of the priority docume application from the International Bure * See the attached detailed Office action for a limit	ents have been received. ents have been received in a riority documents have been eau (PCT Rule 17.2(a)).	Application No n received in this National Stage
Attachment(s)	A) 🗍 Interview	Summary (PTO-413)
 Notice of References Cited (PTO-892) Notice of Draftsperson's Patent Drawing Review (PTO-948) Information Disclosure Statement(s) (PTO-1449 or PTO/SB/0 Paper No(s)/Mail Date 	Paper No	Summary (P10-413) (s)/Mail Date Informal Patent Application (PTO-152)

Art Unit: 1754

DETAILED ACTION

Continued Examination Under 37 CFR 1.114

A request for continued examination under 37 CFR 1.114, including the fee set forth in 37 CFR 1.17(e), was filed in this application after final rejection. Since this application is eligible for continued examination under 37 CFR 1.114, and the fee set forth in 37 CFR 1.17(e) has been timely paid, the finality of the previous Office action has been withdrawn pursuant to 37 CFR 1.114. Applicant's submission filed on 12/20/04 has been entered.

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 1-3, 5, 7-9, and 11-19 are rejected under 35 U.S.C. 103(a) as being unpatentable over Bewersdorf et al. (US 5,560,896) in view of Bewersdorf et al. (USPN 5,714,201).

Bewersdorf ('896) teaches a process for the production of sodium percarbonate. The process comprises spraying an aqueous sodium carbonate solution and a hydrogen peroxide solution onto nuclei in a fluidized bed and evaporating the water. The process utilizes a ternary atomizer nozzle, which allows the solutions to be sprayed through separate channels in order to provide external mixing of the solution, thus preventing the need for condensed phosphates. The hydrogen peroxide solution customarily contains 30-75 % by weight hydrogen peroxide, while

Art Unit: 1754

the sodium carbonate solution contains above 10 %, and especially preferred about 30 %, by weight sodium carbonate. The fluidized bed is maintained at a temperature of between 40 and 95 °C. If needed, additives can be added to either of the hydrogen peroxide and sodium carbonate solutions in order to influence the product qualities and especially to elevate active oxygen stability. The preferred additives are magnesium salts, usually added to the hydrogen peroxide in the form of the sulfate, and water glass, usually added to the sodium carbonate solution.

The amount of the magnesium salt stabilizer is not explicitly taught by Bewersdorf et al., however, it would have been obvious to one of ordinary skill at the time of invention to use an amount between 50 and 2000 ppm, or more specifically between 200 and 1000 ppm, based on the product, as doing so is viewed to be the optimization of a known process, which could have been determined through routine experimentation, and is held to be obvious by *In re Boesch* 205 USPQ 215.

Bewersdorf et al. '896 do not teach the specific amount of waterglass to be added or the module of the waterglass. Bewersdorf et al. '201 teach a similar fluidized bed process with the addition of a sodium silicate with a modulus between 1 and 3 to an aqueous solution. The sodium silicate is introduced in an amount between 0.1 and 2.5 wt%, preferably between 0.5 and 1 wt%, in each case calculated as SiO₂ and relative to sodium percarbonate (column 4, lines 5-11). It would have been obvious to one of ordinary skill in the art at the time of invention to add the sodium silicate, or water glass, as a stabilizer in the process of Brewersdorf et al '896, in the amounts taught by Bewersdorf et al. '201, in order to accomplish the desired effect of the additive.

Art Unit: 1754

Regarding claims 8-11 and 18-19, it is expected that the sodium percarbonate produced by the process of Bewersdorf et al., as above, will have identical properties to those claimed, as they are produced by the same process.

Claims 1-3, 5, 7-9, and 11-19 are rejected under 35 U.S.C. 103(a) as being unpatentable over Bewersdorf et al. ('896) in view of Brichard et al. (US 4,428,914) and further in view of Bewersdorf et al. (USPN 5,714,201).

Bewersdorf teaches a process for the production of sodium percarbonate. The process comprises spraying an aqueous sodium carbonate solution and a hydrogen peroxide solution onto nuclei in a fluidized bed and evaporating the water. The process utilizes a ternary atomizer nozzle, which allows the solutions to be sprayed through separate channels in order to provide external mixing of the solution, thus preventing the need for condensed phosphates. The hydrogen peroxide solution customarily contains 30-75 % by weight hydrogen peroxide, while the sodium carbonate solution contains above 10 %, and especially preferred about 30 %, by weight sodium carbonate. The fluidized bed is maintained at a temperature of between 40 and 95 °C. If needed, additives can be added to either of the hydrogen peroxide and sodium carbonate solutions in order to influence the product qualities and especially to elevate active oxygen stability. The preferred additives are magnesium salts, usually added to the hydrogen peroxide in the form of the sulfate, and water glass, usually added to the sodium carbonate solution. Bewersdorf et al. does not explicitly teach the amounts of these additives.

Brichard, in a similar process, teaches that additives, such as stabilizers of magnesium sulfate and waterglass are added to the aqueous solutions. The stabilizers are generally added in

Art Unit: 1754

amounts between about 1-20 g of stabilizer per kg of sodium percarbonate product, which is equivalent to between 100 - 20,000 ppm (column 4, lines 19-31). It would have been obvious to one of ordinary skill at the time of invention in the process of Bewersdorf et al. to add the stabilizers in an amount corresponding to the amount generally used, as taught by Brichard et al. to accomplish the desired effect of the additives.

Bewersdorf et al. '896 do not teach the specific amount of waterglass to be added or the module of the waterglass. Bewersdorf et al. '201 teach a similar fluidized bed process with the addition of a sodium silicate with a modulus between 1 and 3 to an aqueous solution. The sodium silicate is introduced in an amount between 0.1 and 2.5 wt%, preferably between 0.5 and 1 wt%, in each case calculated as SiO₂ and relative to sodium percarbonate (column 4, lines 5-11). It would have been obvious to one of ordinary skill in the art at the time of invention to add the sodium silicate, or water glass, as a stabilizer in the process of Brewersdorf et al '896, in the amounts taught by Bewersdorf et al. '201, in order to accomplish the desired effect of the additive.

Regarding claims 8-11 and 18-19, it is expected that the sodium percarbonate produced by the process of Bewersdorf et al., as above, will have identical properties to those claimed, as they are produced by the same process.

Claim 4 is rejected under 35 U.S.C. 103(a) as being unpatentable over Bewersdorf et al. ('896) in view of Britchard et al. and further in view of Bewersdorf et al. ('201) or Bewersdorf et al. ('896) in view of Bewersdorf et al. ('201) as applied to claim 1 above, and further in view of Horne et al. (US 6,413,927).

Art Unit: 1754

Bewersdorf et al. teaches that the stabilizing additives may include complexing agents. Bewersdorf et al. does not explicitly teach the list of complexing agents useful as stabilizers. Horne et al. teaches that the percarbonate particles may include a variety of stabilizers, including magnesium salts, alkali silicates (i.e. waterglass), and many of the complexing agents listed in claim 4 (column 6, lines 7-25). It would have been obvious to one of ordinary skill to use a complexing agent from the list of useful stabilizers taught by Horne et al., in the process of Bewersdorf et al., as they meet the desired effect of providing stabilization to the products.

Response to Arguments

Applicant's arguments filed 12/20/04 have been fully considered but they are not persuasive. The applicants argue that Bewersdorf et al. ('201) teaches only the use of waterglass having a specific modulus in the process of coating of sodium percarbonate particles and not in the production of the particles themselves. This is not seen to be the case. Bewersdorf specifically states that the waterglass may be added as stabilizer by fluidized bed spray granulation (abstract) in a manner identical to that claimed by the applicants. Bewersdorf teaches that an alkali metal silicate solution having a modulus in the range from less than 3 to greater than 1, preferably a water glass solution having a modulus in the range from 1.5 to 2.3, may be integrated into conventional processes for the production of sodium percarbonate in a fluidized bed. In known fluidized bed processes, an aqueous hydrogen peroxide solution and an aqueous sodium carbonate solution are sprayed into a fluidized bed dryer with simultaneous vaporization of water. The alkali metal silicate solution is added to the one of the solutions, preferably to the latter. The introduced quantity of alkali metal silicate is in the range between

Application/Control Number: 09/961,395 Page 7

Art Unit: 1754

0.1 and 5 wt. %, preferably from 0.25 to 2.5 wt. %, in each case calculated as SiO₂ and relative to sodium percarbonate. The sodium silicate solution preferably used has a modulus in the range from less than 2 and greater than 1.5, such as a conventional commercial solution having a modulus of 1.8 (column 3, lines 11-35). No difference is seen between the addition of waterglass as a stabilizer in the spray granulation process of Bewersdorf et al. and that of the instantly claimed invention.

Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Peter J Lish whose telephone number is 571-272-1354. The examiner can normally be reached on 9:00-6:00 Monday through Friday.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Stanley Silverman can be reached on 571-272-1358. The fax phone number for the organization where this application or proceeding is assigned is 703-872-9306.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

STANLEY S. SILVERMAN
SUPERVISORY PATENT EXAMINER
TECHNOLOGY CENTER 1700

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